

- The element boron forms a series of hydrides, which includes B_2H_6 , B_4H_{10} , B_5H_9 , B_6H_{10} and $B_{10}H_{14}$. Which one of these hydrides consists of 85.63% boron by mass?

Marks
2

Answer:

- Complete the following table.

2

Formula	Name
K_2SO_4	
	copper(II) chloride
SF_4	
	potassium chromate