

- A mobile phone sends signals at about 850 MHz (1 MHz =  $1 \times 10^6$  Hz). What is the wavelength of this radiation?

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**The frequency,  $\nu = 850$  MHz =  $850 \times 10^6$  Hz, is related to the wavelength,  $\lambda$ , by the equation:**

$$c = \lambda\nu \text{ or } \lambda = \frac{c}{\nu} \text{ where } c \text{ is the speed of light.}$$

$$\text{Therefore, wavelength} = \lambda = \frac{2.998 \times 10^8 \text{ m s}^{-1}}{850 \times 10^6 \text{ s}^{-1}} = 0.35 \text{ m}$$

$$\text{Wavelength} = 0.35 \text{ m}$$