•	Calculate the number of aluminium atoms in a block of pure aluminium that
	measures 2.0 cm $\times$ 2.0 cm $\times$ 3.0 cm. The density of aluminium is 2.7 g cm <sup>-3</sup> .

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$\text{liteasures 2.0 cm} \times 2.0 \text{ cm} \times 3.0 \text{ cm}. \text{ T}$	le density of aluminum is 2.7 g cm <sup>-</sup> .
	Answer: