Explain what is meant by the "common ion effect".	
Magnesium hydroxide is sparingly soludissolution in water and the expression	able. Write down the chemical equation for its for $K_{\rm sp}$ .
What is the molar solubility of magnes	ium hydroxide in water? $K_{\rm sp} = 7.1 \times 10^{-12}$
	<del>*</del>
	Answer:
What is the pH of a saturated solution of	of magnesium hydroxide in water?
	Angreson
	Answer: