• Balance the following nuclear reactions by identifying the missing nuclear particle or nuclide.		Marks 3
$^{63}_{28}\text{Ni} \rightarrow ^{6}_{2}$	³³ ₉ Cu +	
$^{53}_{26}$ Fe +	$^{0}_{-1}e \rightarrow$	
$^{28}_{14}$ Si + $^{2}_{1}$ H -	$\rightarrow {}^{1}_{0}n +$	
• Calculate the energy (in J) and the wavelength (in nm) of the photon of radiation emitted when the electron in Be ³⁺ drops from an $n = 3$ state to an $n = 2$ state.		3
Energy:	Wavelength:	