• Write down the ground state electron configurations for the following species. Na is given as an example.

Marks 4

| Na    | [Ne] $3s^1$ |
|-------|-------------|
| K     |             |
| As    |             |
| Sr    |             |
| $C^+$ |             |

Name the elements described by the following configurations.

[Kr]  $5s^2 4d^6$ 

[Xe]  $6s^2 5d^1 4f^{11}$