CHEM1102	2003-J-4	June 2003
• The solubility product of BaSO ₄ in g L^{-1} ?	constant of BaSO ₄ is $1.1 \times 10^{-10} \text{ M}^2$. Wh	hat is the solubility of 3 Marks 3
	ANSWER:	
• The solubility product constant of Ag_2CrO_4 is 2.6×10^{-12} M ³ . What is the molar solubility of Ag_2CrO_4 in water?		What is the molar 5
	ANSWER:	
What is the molar solubility of Ag_2CrO_4 in a solution of 0.10 M AgNO ₃ ?		AgNO ₃ ?
	ANSWER:	