

Marks
3

- The solubility product constant of BaSO_4 is $1.1 \times 10^{-10} \text{ M}^2$. What is the solubility of BaSO_4 in g L^{-1} ?

ANSWER:

5

- The solubility product constant of Ag_2CrO_4 is $2.6 \times 10^{-12} \text{ M}^3$. What is the molar solubility of Ag_2CrO_4 in water?

ANSWER:

What is the molar solubility of Ag_2CrO_4 in a solution of 0.10 M AgNO_3 ?

ANSWER: