

- The K_{sp} of $\text{Al}(\text{OH})_3$ is $1.0 \times 10^{-33} \text{ M}^4$. What is the solubility of $\text{Al}(\text{OH})_3$ in g L^{-1} ?

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Answer:

What is the solubility of $\text{Al}(\text{OH})_3$ in g L^{-1} at pH 4.00?

Answer: