Marks 8





Assign the stereocentre in compound \mathbf{F} as (*R*) or (*S*), explaining your reasoning.

The priority of the groups around the stereocentre in F is shown above. Br has the highest atomic number and has the highest priority. H has the lowest atomic number and has the lowest priority. It is placed at the back. The other two groups both have carbon bonded to the stereocentre but the group on the left has a higher priority as it has C atoms attached whereas the group on the right has H atoms attached.

With the H atom at the back, the groups are arranged anti-clockwise and so the stereocentre is assigned as (S).



When G is reacted with dilute sulfuric acid, a further product, H, is formed. H has a peak at 3300 cm^{-1} in its IR spectrum. Draw the structure of product H.



stereochemistry.

Is **H** formed as a single enantiomer, as a racemate, or is **H** achiral?

Assuming an S_N^2 mechanism, draw the product of the substitution reaction between **F** and (CH₃)₂NH, indicating stereochemistry where appropriate.