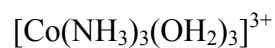
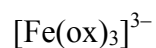
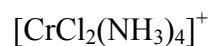


Marks
9

- The following three complex ions can all exhibit isomerism. Name the type of isomerism involved in each case and draw the structures of the isomeric pairs.
ox = oxalate = $\text{C}_2\text{O}_4^{2-}$

**3**

- Give the systematic name of each of the following compounds.
en = ethylenediamine = 1,2-diaminoethane = $\text{NH}_2\text{CH}_2\text{CH}_2\text{NH}_2$

