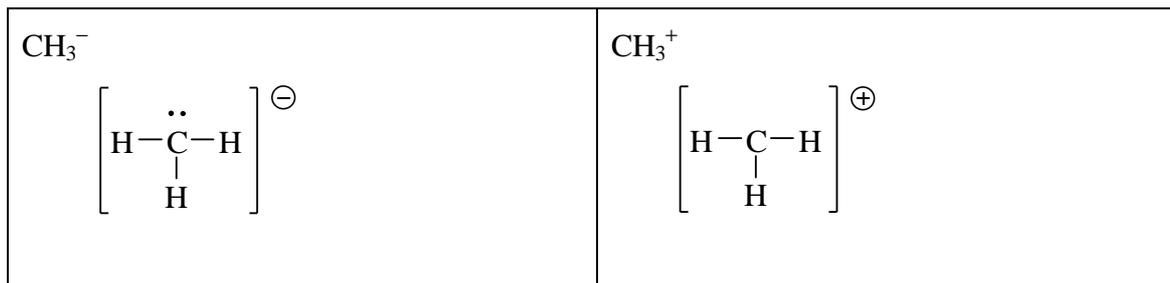


- Draw the Lewis structures, showing all valence electrons for the following species.



Indicate which of these species you expect will be more stable and explain why.

CH_3^- is more stable as it has a full octet of electrons