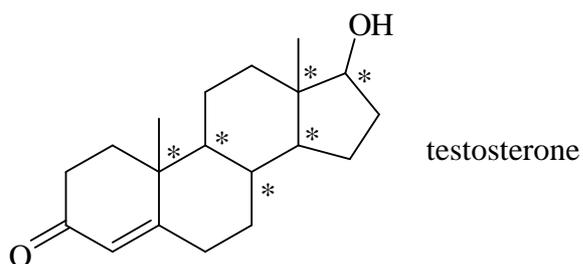


The structure of testosterone, an important male hormone, is shown below.

**Marks**  
**8**



Give the molecular formula of testosterone.



Identify the functional groups present in testosterone.

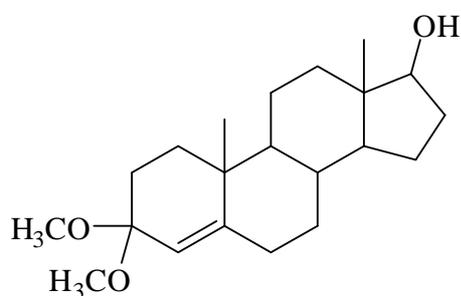
**conjugated ketone, alkene, alcohol (secondary)**

How many stereogenic (chiral) centres are there in testosterone?

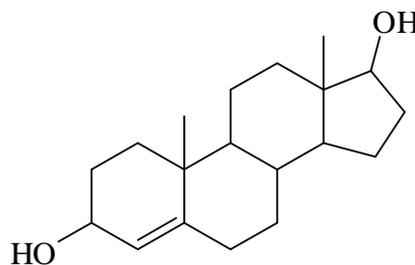
**6 (marked above)**

Draw the constitutional formula of the product formed when testosterone is treated with the following reagents.

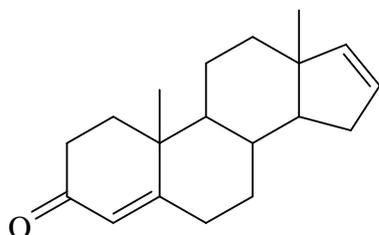
excess methanol / HCl



$\text{LiAlH}_4$  in dry ether; then  $\text{H}^+ / \text{H}_2\text{O}$



concentrated  $\text{H}_2\text{SO}_4$  / heat



$\text{H}_2$  / Pd catalyst

