•	Consider	the	following	reaction.
---	----------	-----	-----------	-----------

 $H_2O(g) + Cl_2O(g)$   $\Longrightarrow$  2HOCl(g)  $K_p = 0.090$  at 298 K Calculate  $\Delta G^{\circ}$  (in J mol<sup>-1</sup>) for this reaction.

$$\Delta G^{\circ} =$$

Calculate the reaction quotient, Q, at 25 °C when  $p(H_2O) = 18$  mmHg,  $p(Cl_2O) = 2.0$  mmHg and p(HOCl) = 0.10 mmHg.

$$Q =$$

In which direction will the reaction proceed spontaneously at these partial pressures?

3