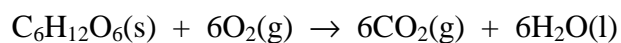


- The balanced equation for the complete oxidation of glucose to carbon dioxide and water is given below.



Calculate the mass of carbon dioxide produced by the complete oxidation of 1.00 g of glucose.

Marks
3

Answer:

Calculate the volume of this mass of carbon dioxide at 0.50 atm pressure and 37 °C.

Answer: