•	Calculate the standard free-energy	change for	or the following	reaction at 298 K.
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Marks 2

Answer:

Complete and balance the following equation for the reaction between iron(II) ions and permanganate ions in an acidic solution.

 $Fe^{2^+} \ + \ MnO_4{}^- \ \rightarrow \ Fe^{3^+} \ + \ Mn^{2^+}$

• What is the value of the equilibrium constant for the following reaction at 298 K?

$$2Fe^{3+}(aq) + Sn(s)$$
 \implies $Sn^{2+}(aq) + 2Fe^{2+}(aq)$

2

2

Answer: