• Order the following salts from lowest to highest molar solubility.

Marks 4

	G. GI			A C O
Salt	CuCl	Cd(IO <sub>3</sub> ) <sub>2</sub>	BaSO <sub>4</sub>	Ag <sub>2</sub> CrO <sub>4</sub>
$K_{ m sp}$	$1.9 \times 10^{-7}$	$2.3 \times 10^{-8}$	$1.1 \times 10^{-10}$	$2.6 \times 10^{-12}$