

- Give the oxidation number of the indicated atom in the following compounds.

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Compound	Atom	Oxidation number
HNO <sub>2</sub>	N	<b>+III or +3</b>
NI <sub>3</sub>	N	<b>-III or -3</b>
[Co(NH <sub>3</sub> ) <sub>5</sub> Cl]SO <sub>4</sub>	Co	<b>+III or +3</b>
K <sub>3</sub> [CrCl <sub>6</sub> ]	Cr	<b>+III or +3</b>