• What is the value of the enthalpy change for the following reaction?

$$MgO(s) + CO_2(g) \rightarrow MgCO_3(s)$$

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Data:	Compound	MgO(s)	$CO_2(g)$	MgCO <sub>3</sub> (s)	
	$\Delta_{\rm f} H^{\circ} /  {\rm kJ}  {\rm mol}^{-1}$	-602	-394	-1096	
		F			
			Answer:		

THE REMAINDER OF THIS PAGE IS FOR ROUGH WORKING ONLY.