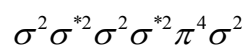


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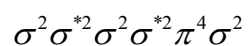
- The electronic configuration of molecular nitrogen in its ground state is, in order (from left to right) of orbitals of increasing energy:



What is the bond order of N₂?

How many of the valence electrons in N₂ are in non-bonding 'lone pairs' according to Lewis theory?

On the electron configuration of N₂ below, indicate by arrows the molecular orbitals that contain the non-bonding electrons.



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