• Explain why a sustained fission chain reaction can only occur when a critical mass is prepared.

Marks 2

• The half life of ³H is 12 years. Calculate how long it takes (rounded to the nearest year) for the activity of a sample of tritium to have dropped to 0.1% of its original value.

2

Answer:

• Consider the following list of unstable isotopes and their decay mechanisms.

3

$$^{33}_{17}\text{Cl} \ \to \ ^{0}_{+1}\text{e} \ + \ ^{33}_{16}\text{S}$$

half-life =
$$2.5 \text{ s}$$

$$^{32}_{15}P$$
 \rightarrow $^{0}_{-1}e$ + $^{32}_{16}S$ half-life = 14.3 days

$$^{199}_{82}\text{Pb}$$
 \rightarrow $^{0}_{+1}\text{e}$ + $^{199}_{81}\text{Tl}$ half-life = 90 minutes

$$^{13}_{7}N \rightarrow ^{0}_{+1}e + ^{13}_{6}C$$

half-life = 10 minutes

From this list, select the isotope that best satisfies the following requirements. Provide a reason for your choice in each case.

Requirement	Isotope	Reason for choice
Isotope used in medical imaging		
Decay represents the transformation of a neutron into a proton		
The isotope with the highest molar activity		