1. Which of the curly arrows are incorrectly drawn in the following reaction mechanism?
a) $\mathbf{X}$ only
b) $\mathbf{Y}$ only
c) $\mathbf{Z}$ only
d) all of them

e) none of them\#
2. Which one of the following reagents would best effect the conversion shown?
a) $\mathrm{H}_{2} / \mathrm{Pd}$ catalyst
b) $\mathrm{NaBH}_{4}$ followed by $\mathrm{H}^{+} / \mathrm{H}_{2} \mathrm{O}$
c) $\mathrm{Cr}_{2} \mathrm{O}_{7}^{2-} / \mathrm{H}^{+}$
d) hot conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$

e) hot dilute $\mathrm{OH}^{-}$
3. Which one of the following reagents would best effect the conversion shown?
a) conc. HCl
b) excess $\mathrm{NH}_{3}$
c) $\mathrm{SOCl}_{2}$
d) $\mathrm{H}^{+} / \mathrm{H}_{2} \mathrm{O} /$ heat

e) $\mathrm{OH}^{-} / \mathrm{H}_{2} \mathrm{O} /$ heat
4. What are the major products from the hydrolysis of $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{COOCH}_{3}$ with hot 6 M HCl ?
a) $P$ and $R$
b) $P$ and $S$
c) $Q$ and $R$
d) Q and S
$\mathrm{CH}_{3} \mathrm{OH}$
P

Q

R

S
5. Which definition best describes the following pair of compounds?
a) Enantiomers
b) Diastereomers
c) Constitutional isomers
d) Conformers


e) Same compound
6. What is the order of priority ( $1^{\text {st }}=$ highest $)$ and the absolute configuration of the following compound?

|  | $1^{\text {st }}$ | $2^{\text {nd }}$ | $3^{\text {rd }}$ | $4^{\text {th }}$ | Abs. Config. |
| :--- | :--- | :--- | :--- | :---: | :--- |
| a) Br | $\mathrm{CH}_{3}$ | COOH | H | $(S)$ |  |
| b) Br | COOH | $\mathrm{CH}_{3}$ | H | $(R)$ |  |
| c) COOH | Br | $\mathrm{CH}_{3}$ | H | $(R)$ | COOH |
| d) Br | COOH | $\mathrm{CH}_{3}$ | H | $(S)$ | $\mathrm{Br}_{3}$ |
| e) COOH | Br | $\mathrm{CH}_{3}$ | H | $(S)$ |  |

7. Which of the following terms best describes the product from the following reaction?
a) ( $R$ )-enantiomer
b) (S)-enantiomer
c) racemic mixture
d) achiral compound
e) meso-compound

$(R)$-enantiomer
8. The reagents and reaction conditions to carry out the transformations below are:

a) (i) hot dilute $\mathrm{H}_{2} \mathrm{SO}_{4}$ (ii) $\mathrm{Cr}_{2} \mathrm{O}_{7}^{2-} / \mathrm{H}^{+}$(iii) excess ethanol/ $/ \mathrm{H}^{+}$
b) (i) hot dilute $\mathrm{OH}^{-} / \mathrm{H}_{2} \mathrm{O}$ (ii) $\mathrm{Cr}_{2} \mathrm{O}_{7}^{2-} / \mathrm{H}^{+}$(iii) $\mathrm{NaOCH}_{2} \mathrm{CH}_{3}$ in ethanol
c) (i) hot dilute $\mathrm{H}_{2} \mathrm{SO}_{4}$ (ii) $\mathrm{OH}^{-} / \mathrm{H}_{2} \mathrm{O}$ (iii) excess ethanol
d) (i) hot conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$ (ii) $\mathrm{LiAlH}_{4}$ followed by dilute acid (iii) $\mathrm{NaOCH}_{2} \mathrm{CH}_{3}$ in ethanol
e) (i) hot dilute $\mathrm{OH}^{-} / \mathrm{H}_{2} \mathrm{O}$ (ii) $\mathrm{Cr}_{2} \mathrm{O}_{7}^{2-} / \mathrm{H}^{+}$(iii) excess ethanol $/ \mathrm{H}^{+}$

Q 9-10 refer to the following four molecules.

tyramine

tropidene

vanillin

morpholine
9. Which of the compounds will undergo an acid-base reaction with dilute HCl ?
a) tyramine and tropidene only
d) tropidene only
e) all of them
10. Which of the compounds will undergo an acid-base reaction with dilute NaOH ?
a) tyramine and vanillin only
b) tyramine, vanillin and morpholine only
c) tyramine only
d) vanillin only
e) none of them

Correct answers: $\quad 1 \mathrm{E}, 2 \mathrm{~B}, 3 \mathrm{E}, 4 \mathrm{~A}, 5 \mathrm{C}, 6 \mathrm{D}, 7 \mathrm{D}, 8 \mathrm{~A}, 9 \mathrm{~B}, 10 \mathrm{~A}$

